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PHOTOCHEMICAL SYNTHESIS OF 6-ARYLURIDINES¹⁾

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Abstract: Synthesis of 6-aryluridines was effected by photochemical arylation of 6-iodo-2',3'-0-iso-propylidene-5'-0-methoxymethyluridine.

Our recent publications on the lithiation of nucleosides $^{2-7)}$ have proved the effectiveness of this tactic for the modification of base moieties both in terms of simplicity and generality. Thus, various types of 6-substituted uridines were prepared by the metallation of 2',3'-0-iso-propylidene-5'-0-methoxymethyluridine (1) with lithium diisopropylamide (LDA) and successive reactions with electrophiles. 2

Some limitations involved in the above reaction can be compensated by a nucleophilic addition-elimination reaction of a compound having a suitable leaving group at the C-6 position, as has been shown in the preparation of 6-azido-and 6-alkylthiouridines. 8,9)

With an intention to synthesize 6-aryluridines, we initially examined the addition-elimination reaction of $\underline{2}$ - $\underline{4}$ by using phenyllithium as a nucleophile. However, the main product in these reactions was $\underline{1}$. 6-Phenyluridine derivative $\underline{5}$ was isolated only in the reaction of $\underline{2}$ (3.5 eq of PhLi, in THF,

1 X= H 2 X= SPh 3 X= SePh 4 X= I 5 X= Ph

below -70 °C, 3 h), but the yield of $\underline{5}$ was low (18%). This

prompted us to investigate the photochemical synthesis of 6-aryluridines from 6-iodo-2',3'-0-isopropylidene-5'-0-methoxymethyluridine ($\underline{4}$). While this work was in progress, a brief communication appeared in which a photochemical reaction of $\underline{4}$ prepared by our method with \underline{N} -phenylpyrrole has been described. We are not aware of any other report on the arylation of uridine at the C-6 position.

When $\underline{4}$ was irradiated in benzene for 9 h with a 400 W high-pressure Hg lamp equipped with a Pyrex filter, $\underline{5}$ was produced in 12% yield. Another product isolated in this reaction showed H-6 (δ 7.32 ppm, doublet), which was coupled

with H-5 (δ 5.78 ppm, double doublet), in its PMR spectrum measured in CDCl₃. In addition, it was void of a signal corresponding to H-4'. On the basis of these PMR data and those reported for 1-(5-deoxy-2,3-Q-isopropylidene- β -D-erythro-pent-4-enofuranosyl)uracil, the structure of this product was determined as $\underline{6}$ (8%). Its MS spectrum (m/z 327: M+1, m/z 311: M-Me, m/z 112: B+1) was also in accord with this structure.

Addition of Et_3N (1 mol eq to $\underline{4}$) in the above phenylation gave a higher yield of $\underline{5}$ (43%) and the formation of $\underline{6}$ was not observed in this case. When the reaction was carried out in the presence of Et_3N by using MeCN as a co-solvent, the reaction time required for disappearance of the starting material was reduced to 4 h, but $\underline{5}$ was isolated only in 5% yield together with 1 (26%).

Trapping of the C-6 radical generated from $\underline{4}$ was further examined with other solvents in the absence or presence of MeCN. These results are summarized in Tables 1 and 2. As

Table 1 Synthesis of 6-Aryluridines in the presence of 1 eq of Et₇N

the presence of 1 eq of Etzn			
React. Time(h)	Product	Isolated yield(%)	
9	<u>5</u>	43	
3	<u>7</u>	54	
7	8	20	
le 13	9	62	
1 3	<u>10</u>	18	
	React. Time(h) 9 3 7 1e 13	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	

Solvent	MeCN/Solvent*	React.Time(h) Products [†]
benzene	1	4	5(5%), $1(26%)$
anisole	1	8	7(18%), $1(14%)$
thiophene	3	3	8(10%), $1(9%)$, $4(4%)$
\underline{N} -methylpyrr	role 3	3	$9(13\%), \underline{4}(28\%)$

^{*} ratio in v/v
† isolated yields are shown
in parenthese

can be seen from these results, higher yields of 6-arylated products were obtained when the reactions were carried out in the absence of MeCN.

From the PMR spectra of 8, 9, and 10, it became clear that all the reactions with heteroaromatic solvents had occurred at the α -position. In the reaction with anisole, the PMR spectrum of the product (7) indicated the presence of two regioisomers, which could not be separated even after deprotection or by the successive acetylation.

Another PMR feature of these products deserves a short comment, since introduction of an aryl group to the C-6 position decreased the differences in the chemical shift between the isopropylidene methyl signals, $\Delta\delta$ Me value, ¹⁵⁾ as shown in Table 3.

We have already observed that a 6-substituted uridine having a "C⁶-C-Ar" structure, where the aryl ring can bend to the <u>endo</u> isopropylidene methyl group, shows a considerably reduced $\Delta\delta$ Me value. However, the above PMR results of 6-aryluridine derivatives suggest that, quite unexpectedly from examination of a molecular model, even a "C⁶-Ar" structure in the uracil moiety can cause a change of the $\Delta\delta$ Me value.

In conclusion, the usefulness of a 6-iodouridine derivative was exemplified herein by its photochemical arylation. 5-Substituted 6-arylaridines can certainly be pre-

protonated species.

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Table 3	PMR chemical shifts (δ: pp	m) of
	isopropylidene Me signals	in CDC1 ₃

endo-Me	exo-Me	Δδ Me (ppm)
1.59	1.36	0.23
1.57	1.35	0.22
1.38	1.30	0.08
1.45	1.32	0.13
1.45	1.32	0.13
1.51	1.35	0.16
	1.59 1.57 1.38 1.45	1.59 1.36 1.57 1.35 1.38 1.30 1.45 1.32 1.45 1.32

pared in a similar manner, since their 6-iodo derivatives are now easily accessible by the lithiation method. $^{17})$

EXPERIMENTAL

Melting points were determined with a Yanagimoto micromelting point apparatus and are uncorrected. PMR spectra were measured with a JEOL JNM-FX 100 NMR spectrometer by using TMS (tetramethylsilane) as an internal standard. The abbreviations used are as follows: s, singlet; d, doublet; dd, double doublet; t, triplet; m, multiplet; br, broad. Mass spectra were taken on a JEOL JMS-D 300 spectrometer. UV spectra were recorded on a Shimadzu UV-240 spectrophotometer. Column chromatography was carried out on silica gel (Wakogel® C-200). TLC was performed on silica gel (precoated silica gel plate F_{254} , Merck).

General procedure for photoarylation of 4 — Irradiation of $\underline{4}$ (200 mg) was carried out, with a Shigemi 400 W high-pressure mercury lamp equipped with a Pyrex filter, in an appropriate solvent (80 ml) and $\mathrm{Et}_3\mathrm{N}$ (0.06 ml) under Ar atmosphere. After evaporation of the solvent, each product was purified by preparative TLC.

2',3'-0-Isopropylidene-5'-0-methoxymethyl-6-phenyluridine (5)— This compound was obtained as colorless oil.

PMR (CDC1₃) δ : 1.30 (3H, s, isop.Me), 1.38 (3H, s, isop.Me), 3.37 (3H, s, CH₂OCH₃), 3.79 (2H, m, CH₂-5'), 4.16 (1H, m, H-4'), 4.68 (2H, s, CH₂OCH₃), 4.86 (1H, dd, H-3'), 5.23 (1H, dd, H-2'), 5.52 (1H, d, J= 1.5 Hz, H-1'), 5.65 (1H, s, H-5), 7.50 (5H, s, pheny1), 9.49 (1H, br, NH). MS m/z: 404 (M⁺), 389 (M-Me), 188 (B+1).

- $\frac{1-(2,3-0-Isopropylidene-5-0-methoxymethyl-8-D-erythro-pent-4-enofuranosyl)uracil (6)}{\text{pent-4-enofuranosyl})uracil (6)}$ This compound was obtained as a foam. PMR (CDCl₃) δ : 1.38 (3H, s, isop.Me), 1.55 (3H, s, isop.Me), 3.49 (3H, s, CH₂OCH₃), 4.76 (1H, d, J= 1.5 Hz, H-5'), 5.19 (1H, d, J= 6.3 Hz, H-2'), 5.29 (2H, s, CH₂OCH₃), 5.41 (1H, dd, H-3'), 5.53 (1H; s, H-1'), 5.78 (1H, dd, H-5), 7.32 (1H, d, H-6), 8.73 (1H, br, NH). MS m/z: 327 (M+1), 311 (M-Me), 112 (B+1).
- 2',3'-O-Isopropylidene-5'-O-methoxymethyl-6-(methoxy-phenyl)uridine (7)—This compound, a mixture of two regio-isomers, was obtained as foam. Partial PMR data which show the presence of two regioisomers are given below. PMR (CDCl₃) δ : 3.34 and 3.38 (3H, each as s, CH₂OCH₃), 3.82 and 3.88 (3H, each as s, OMe), 4.64 and 4.68 (2H, each as s, CH₂OCH₃), 9.43 and 9.55 (1H, each as br, NH).
- $\frac{2',3'-0-Isopropylidene-5'-0-methoxymethyl-6-(N-methyl-pyrrol-2-yl)uridine (9)}{\text{Down}}$ This compound was obtained as a foam. PMR (CDCl₃) δ : 1.32 (3H, s, isop.Me), 1.45 (3H, s, isop.Me), 3.36 (3H, s, CH₂OCH₃), 3.63 (3H, s, N-Me), 3.79

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(2H, m, CH_2 -5'), 4.15 (1H, m, H-4'), 4.66 (2H, s, CH_2 OCH₃), 4.86 (1H, dd, H-3'), 5.22 (1H, dd, H-2'), 5.56 (1H, d, J= 1.5 Hz, H-1'), 5.66 (1H, d, H-5), 6.23, 6.45, and 6.81 (3H, each as dd, Ar), 9.94 (1H, br, NH). MS m/z: 407 (M⁺), 392 (M-Me), 191 (B+1). High resolution MS m/z: 407.1695 (M⁺) Calcd. for $C_{19}H_{25}N_3O_7$ 407.1683.

6-Phenyluridine (11) — Compound 5 (100 mg) in 50% TFA (5 ml) was stirred for 48 h at room temperature. Evaporation of the solvent followed by short-column chromatography on silica gel(5% EtOH in CHCl₃) gave $\underline{11}$ (50 mg, 63%) as a foam. PMR (CD₃OD) δ : 3.73 (3H, m, CH₂-5' and H-4'), 4.29 (1H, dd, H-3'), 4.74 (1H, dd, H-2'), 5.25 (1H, d, J= 4.0 Hz, H-1'), 5.58 (1H, s, H-5), 7.53 (5H, s, Ph).

Compound $\underline{11}$ was converted to its triacetate, whose high resolution MS was measured. High resolution MS m/z: 446.1372 (M⁺) Ca1cd. for C $_{21}$ H $_{22}$ N $_{2}$ O $_{9}$ 446.1317. UV absorption in MeOH: max 265 nm, min 238 nm.

<u>6-Methoxyphenyluridine (12)</u>—Compound 7 (180 mg) in 50% TFA (5 ml) was stirred for 17 h at room temperature. Evaporation of the solvent followed by short-column chromatography on silica gel (5% EtOH in CHCl₃) gave 12 (121 mg) as colorless oil which was a mixture of two regionsomers.

Compound $\underline{12}$ was converted to its triacetate, whose high resolution MS was measured. High resolution MS m/z: 476.1469 (M⁺) Calcd. for $C_{22}H_{24}N_2O_{10}$ 476.1422.

6-(Thien-2-y1)uridine (13)— Compound 8 (70 mg) in 50% TFA (3 m1) was stirred for 4 h at room temperature. Evaporation of the solvent followed by short-column chromatography on silica gel (5% EtOH in CHCl₃) gave 13 (35 mg, 60%). Crystallization from acetone-hexane gave an analytical sample (mp 111-114 °C). Anal. Calcd. for $C_{13}H_{14}N_{2}O_{6}S \cdot 1/3$ acetone: C, 48.64; H, 4.67; N, 8.11. Found: C, 48.75; H, 4.67; N, 8.16. UV absorption in MeOH: max 292 nm (ε 9300), min 247.5 nm (ε 6400). PMR (CD₃OD) δ : 2.15 (2H, s, acetone), 3.76 (3H, m, CH₂-5' and H-4'), 4.31 (1H, dd, H-3'), 4.76 (1H, dd, H-2'), 5.55 (1H, d, J= 3.5 Hz, H-1'), 5.75 (1H, s, H-5), 7.19, 7.54, and 7.69 (3H, each as dd, Ar).

 $\frac{6-(5-\text{Methylfuran-}2-\text{yl})\text{uridine}(14)}{\text{compound}} = \frac{10}{10} \text{ (40 mg)}$ in 50% TFA (2 ml) was stirred for 16 h at room temperature. Evaporation of the solvent followed by short-column chromatography on silica gel (5% EtOH in CHCl₃) gave $\frac{14}{10} \text{ (23 mg, 73\%)}$. Crystallization from acetone-hexane gave an analytical sample (mp 282-287 °C). Anal. Calcd. for $C_{14}H_{16}N_{2}O_{7}$: C, 51.84; H, 4.98; N, 8.64. Found: C, 52.11; H, 5.12; N, 8.59. UV absorption in MeOH: max 315 nm (ε 15000), shoulder 281 nm (ε 7500) and 271 nm (ε 7100), min 241.5 nm (ε 2500). PMR (CD₃OD) δ : 2.39 (3H, d, J= 1.0 Hz, Ar-Me), 3.78 (3H, m, CH₂-5' and H-4'), 4.34 (1H, dd, H-3'), 4.78 (1H, dd, H-2'), 5.71 (1H, d, J= 3.5 Hz, H-1'), 5.87 (1H, s, H-5), 6.28 (1H, dd, Ar), 7.10 (1H, d, Ar).

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